

la masse atomique moyenne

✓ #1 p. 167
 $19,78\% \text{ } ^{10}_5\text{B} (10,01\text{u})$

$80,22\% \text{ } ^{11}_5\text{B} (11,01\text{u})$

$$0,1978 (10,01\text{u}) + 0,8022 (11,01\text{u}) = 10,81\text{u}$$

✓ 2. $0,9223 (27,98\text{u}) + 0,10467 (28,97\text{u}) + 0,0310 (29,97\text{u}) = 28,09\text{u}$

✓ 3. $62,93\text{u} (0,691) + 64,93\text{u} (0,309) = 63,6\text{u}$

✓ 4. $0,0137 \times 2040\text{u} + 0,2626 (206,0\text{u}) + 0,2082 (207,0) +$
 $0,5155 (208,0\text{u}) = 207,2\text{u}$